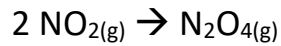


ICE Table Practice Problem #2

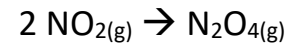
In the following reaction, $K_{eq} = 9.3 \times 10^{-7}$ at room temp. Calculate the equilibrium concentration of N_2O_4 in a flask initially containing only 3.00 M of NO_2



Rxn	2 NO _{2(g)}	→	N ₂ O _{4(g)}
I			
C			
E			
5%			
Answer			

ICE Table Practice Problem #2

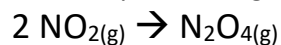
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ICE Table Practice Problem #2

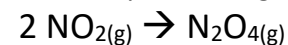
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