

ICE Table Practice Problem #1

If you have an initial concentration of $[\text{PCl}_5]$ at 1.3M, what are the concentrations of the products at equilibrium? Assume all reactants and products are aqueous and $K_{\text{eq}} = 78.3$.



Rxn	$[\text{PCl}_5]$	$[\text{PCl}_3]$	$[\text{Cl}_2]$
I			
C			
E			
5%			
Answer			

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