

Things to look for BEFORE answering an equilibrium Q		
Stressor	Question	What does it tell us?
Increase or decrease [ ] products or reactants	Which phase?	<ul style="list-style-type: none"> <li>Gas, aqueous - change things</li> <li>Solid, Liquid – DON'T CHANGE ANYTHING!</li> </ul>
Increase or decrease T	Endo or exo?	<ul style="list-style-type: none"> <li>Endo = absorbed, so it is a REACTANT</li> <li>Exo = released, so it is a PRODUCT</li> </ul>
Increase or decrease total Pressure <i>(Same as <math>\Delta</math> in V or # of moles of gas)</i>	How many moles of GAS are on each side of the equation?	<ul style="list-style-type: none"> <li>Increase pressure = move to side with FEWER moles of gas</li> <li>Decrease pressure = move to side with MORE moles of gas</li> </ul>

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