

$\text{pH} = -\log [\text{H}^+]$	$\text{pOH} = -\log [\text{OH}^-]$
$[\text{H}^+] = 10^{-\text{pH}}$	$[\text{OH}^-] = 10^{-\text{pOH}}$
$\text{pH} + \text{pOH} = 14$	
$[\text{H}^+][\text{OH}^-] = 1 \times 10^{-14}$	

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