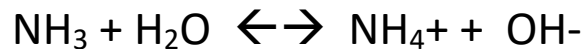


Weak Acid/Base Practice Problem #2

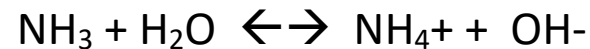
You have 0.010 M NH_3 . Calculate the pH. $K_b = 1.8 \times 10^{-5}$



Rxn	NH_3	\leftrightarrow	NH_4^+	+	OH^-
I					
C					
E					
5%					
Answer					

Weak Acid/Base Practice Problem #2

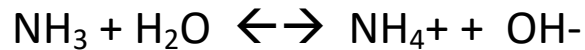
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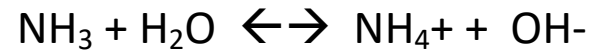
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